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Section 15-5 Spontaneous Processes A spontaneous process is a physical or chemical change that once begun, occurs with no outside intervention. Many spontaneous processes require some energy from the surroundings to start the process. 53.

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Figure 15.4.1 Energy Changes Accompanying the Thermite Reaction Because enthalpy is a state function, the overall enthalpy change for the reaction of 2 mol of Al(s) with 1 mol of Fe 2 O 3 (s) is –851.1 kJ, whether the reaction occurs in a single step (ΔH 4, shown on the left) or in three hypothetical steps (shown on the right) that involve the successive formation of solid Al 2 O 3 and ...

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